

Katarzyna Baranowska, Jarosław Chojnacki,\* Barbara Becker and Wiesław Wojnowski

Department of Chemistry, Technical University of Gdańsk 11/12, G. Narutowicz St., 80952-PL Gdańsk, Poland

Correspondence e-mail:  
jarekch@chem.pg.gda.pl

## Key indicators

Single-crystal X-ray study  
 $T = 200\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.008\text{ \AA}$   
 $R$  factor = 0.039  
 $wR$  factor = 0.144  
Data-to-parameter ratio = 9.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

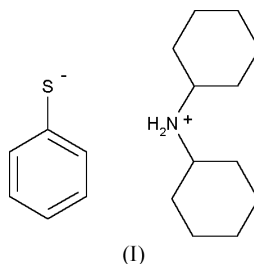
## Dicyclohexylammonium benzenethiolate

The title compound,  $\text{C}_{12}\text{H}_{24}\text{N}^+\cdot\text{C}_6\text{H}_5\text{S}^-$ , has been obtained by the reaction of  $\text{C}_6\text{H}_5\text{SH}$  with dicyclohexylamine. The two ionic fragments are linked together in long chains by  $\cdots\text{N}-\text{H}\cdots\text{S}\cdots$  hydrogen bonds. There are two formula units in the asymmetric unit.

Received 16 April 2003  
Accepted 25 April 2003  
Online 9 May 2003

## Comment

The title compound, (I), has been obtained previously in polycrystalline form (Dance, 1979). It was found to be a good isolable source of the benzenethiolate anion for the preparation of metal thiolates. In addition, there is a need for information in the study of  $\text{N}-\text{H}\cdots\text{S}$  bonding. Such bonding plays a crucial role in biological systems, *e.g.* in the modification of the redox potential in iron-sulfur proteins (Nakamura & Ueyama, 1988). Detailed research on  $\text{N}-\text{H}\cdots\text{S}$  hydrogen bonding in simple molecules is, therefore, important.



The asymmetric unit of (I) consists of two benzenethiolate anions and two dicyclohexylammonium cations. The ions are linked in zigzag chains by  $\text{S}\cdots\text{H}-\text{N}-\text{H}\cdots$  hydrogen bonding, motif  $\text{C}_2^1(4)$  (Etter, 1990). The donor-acceptor  $\text{N}\cdots\text{S}$  distances

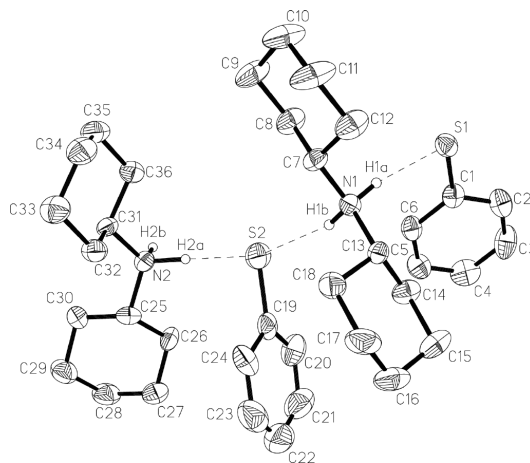
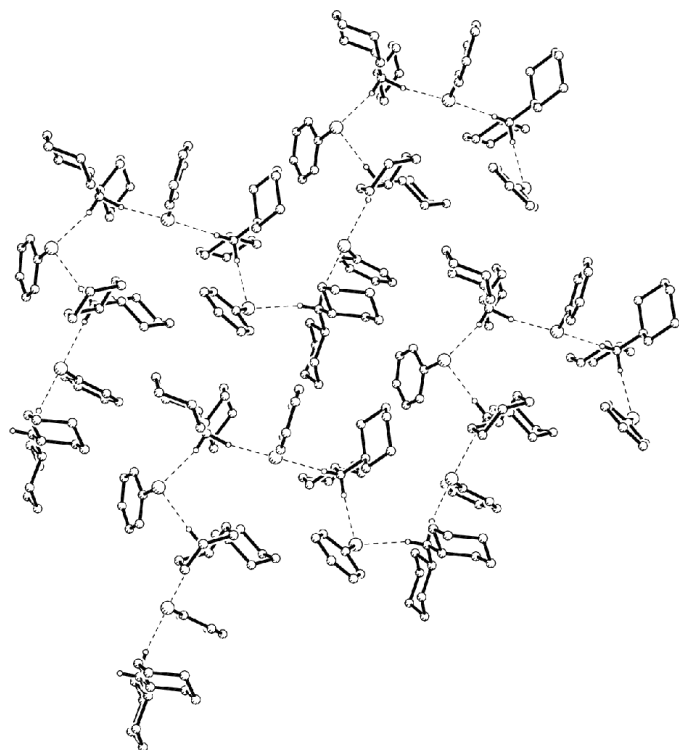


Figure 1

The asymmetric unit of (I), with the atomic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. All H atoms bonded to C atoms have been omitted.



**Figure 2**  
Diagram of the hydrogen-bond network.

[range 3.194 (5)–3.266 (5) Å] are comparable with the values observed in zinc and cobalt benzenethiolate complexes, [(C<sub>6</sub>H<sub>11</sub>)<sub>2</sub>NH<sub>2</sub>]<sub>2</sub>[Zn<sub>2</sub>(SC<sub>6</sub>H<sub>5</sub>)<sub>6</sub>] (Chung *et al.*, 1991a) and [(C<sub>6</sub>H<sub>11</sub>)<sub>2</sub>NH<sub>2</sub>]<sub>2</sub>[Co(SC<sub>6</sub>H<sub>5</sub>)<sub>4</sub>] (Chung *et al.*, 1991b), and in [(CH<sub>3</sub>)<sub>3</sub>NCH<sub>2</sub>CONH<sub>2</sub>][SC<sub>6</sub>H<sub>5</sub>] (Walters *et al.*, 1991).

## Experimental

All procedures were carried out under argon, using standard Schlenk techniques. The solvent and the amine were dried by standard methods, and distilled under argon prior to use. To a solution of benzenethiol (0.22 g, 2 mmol) in acetonitrile, freshly prepared dicyclohexylamine (0.36 g, 2 mmol) was added. Immediately a white precipitate formed. Recrystallization from hot acetonitrile afforded colourless crystals of (I) suitable for X-ray diffraction analysis.

### Crystal data

C <sub>12</sub> H <sub>24</sub> N <sup>+</sup> ·C <sub>6</sub> H <sub>5</sub> S <sup>-</sup>	Mo K $\alpha$ radiation
<i>M<sub>r</sub></i> = 291.50	Cell parameters from 40 reflections
Orthorhombic, <i>Pca</i> 2 <sub>1</sub>	$\theta$ = 2.8–18.7°
<i>a</i> = 21.208 (4) Å	$\mu$ = 0.18 mm <sup>-1</sup>
<i>b</i> = 11.678 (2) Å	<i>T</i> = 200 (2) K
<i>c</i> = 13.974 (3) Å	Plate, colourless
<i>V</i> = 3460.9 (12) Å <sup>3</sup>	0.5 × 0.4 × 0.1 mm
<i>Z</i> = 8	
<i>D<sub>x</sub></i> = 1.119 Mg m <sup>-3</sup>	

### Data collection

Oxford Diffraction KM-4 diffractometer	<i>h</i> = -26 → 0
<i>w</i> -2 $\theta$ scans	<i>k</i> = -14 → 0
3593 measured reflections	<i>l</i> = 0 → 17
3593 independent reflections	3 standard reflections every 200 reflections
2159 reflections with <i>I</i> > 2 $\sigma$ ( <i>I</i> )	intensity decay: 0.3%
$\theta_{\max}$ = 26.4°	

### Refinement

Refinement on <i>F</i> <sup>2</sup>	$w = 1/[\sigma^2(F_o^2) + (0.079P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.039$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.144$	( $\Delta/\sigma$ ) <sub>max</sub> = 0.032
<i>S</i> = 1.04	$\Delta\rho_{\max} = 0.30 \text{ e \AA}^{-3}$
3593 reflections	$\Delta\rho_{\min} = -0.26 \text{ e \AA}^{-3}$
373 parameters	Absolute structure: Flack (1983),
H atoms treated by a mixture of independent and constrained refinement	33 Friedel pairs
	Flack parameter = -0.06 (13)

**Table 1**

Selected geometric parameters (Å, °).

S1—C1	1.762 (5)	N1—C7	1.497 (6)
S2—C19	1.755 (6)	N2—C25	1.490 (6)
N1—C13	1.498 (6)	N2—C31	1.507 (6)
C13—N1—C7	117.6 (4)	C13—N1—H1B	103 (3)
C13—N1—H1A	113 (3)	C7—N1—H1B	113 (3)
C7—N1—H1A	102 (3)	H1A—N1—H1B	109 (4)

**Table 2**

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1A...S1 <sup>i</sup>	0.92 (5)	2.29 (6)	3.200 (4)	170 (4)
N1—H1B...S2	0.93 (6)	2.37 (6)	3.266 (5)	163 (4)
N2—H2B...S1 <sup>ii</sup>	0.87 (6)	2.34 (6)	3.194 (5)	167 (5)
N2—H2A...S2 <sup>iii</sup>	1.11 (5)	2.15 (6)	3.246 (5)	166 (4)

Symmetry codes: (i)  $-x, 2 - y, \frac{1}{2} + z$ ; (ii)  $-x, 1 - y, \frac{1}{2} + z$ ; (iii)  $-x, 1 - y, z - \frac{1}{2}$ .

H atoms bonded to C atoms were treated as riding, with fixed isotropic displacement parameters. The coordinates of the H atoms bonded to N were refined.

Data collection: *KM-4 System* (Gałdecki, Kowalski *et al.*, 1996); cell refinement: *KM-4 System*; data reduction: *DATAPROC* (Gałdecki, Kowalski & Uszyński, 1996); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *RESVIEW* (Schwenk, 1998); software used to prepare material for publication: *SHELXL97*.

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